

Updated 07/18/2023



WACO, TEXAS

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**COURSE SYLLABUS  
AND  
INSTRUCTOR PLAN**

**General Chemistry I**

**CHEM 1411-005**

**Raymond Kessler**

**NOTE: This is a 16-week course.**

**NOTE: This is a Face-to-Face course.**

**Course Description:**

Fundamental principles of chemistry for majors in the sciences, health sciences, and engineering; topics include measurements, fundamental properties of matter, states of matter, chemical reactions, chemical stoichiometry, periodicity of elemental properties, atomic structure, chemical bonding, molecular structure, solutions, properties of gases, and an introduction to thermodynamics and descriptive chemistry. Includes basic laboratory experiments supporting theoretical principles; introduction of the scientific method, experimental design, data collection and analysis, and preparation of laboratory reports.

**Prerequisites and/or Corequisites:**

Prerequisites: MATH 1314 with a minimum grade of C, or passing score on non-credit equivalency exam for MATH 1314, or consent of the division chair. High school chemistry is strongly recommended. Semester Hours 4 (3 lec/3 lab).

**Course Notes and Instructor Recommendations:**

The syllabus, class calendar, lecture materials, quizzes, grades, and other information will be posted on D2L Brightspace. Additionally, it is important to check D2LBrightspace periodically for new assignments and/or announcements. If you have problems with either your computer hardware or software, it is still your responsibility to make sure that all assignments are turned in on time. Extensions of due dates will NOT be given due to failure of computer or internet access. If you cannot access D2L Brightspace from home, MCC has available computers across campus.

This is not a blended/hybrid course; thus, students are expected to be available in person for the class time scheduled. It is recommended that the student attend class regularly, take adequate lecture notes, and study outside of class. Problems sets will require the student be knowledgeable of algebraic manipulation of symbolic equations. Homework sets will help practicing calculations and assist in understanding the concepts covered in each chapter. It is vital that you read the text BEFORE the lecture on that chapter. This will greatly improve your understanding of the topic. Students are expected to spend between 3- 6 hours per week reading, watching screencast lectures, and videos. A general recommendation is 2-3 hours per week of study time for every college credit hour taken.

Dedicate enough time to this course. Waiting until the assignment due date to start the assignment is not a recommended strategy for success. If you have questions, there is not enough time for me to respond before the assignment is due. It is strongly recommended that you start the assignments in advance. Listen and take notes of instructor lectures (avoid distractions)

and/or videos posted by the instructor. Review the PowerPoint presentations provided by the instructor and read the book for further clarification before attempting the assignment.

Contact your instructor if you have any questions about the course, or questions about the material covered. Contact your instructor immediately if you have any problems (personal, technical, etc.) that prevent you from completing your class responsibilities by the deadline. Please use your MCC e-mail account when contacting me. Include a brief description of the purpose of your email. Include your whole name and the course name in the subject line. A student's McLennan email address will be used for official college information or business. Students are expected to read and, if needed, respond in a timely manner to college emails. For more information about your student email account, go to [www.mclennan.edu/student-email](http://www.mclennan.edu/student-email).

**Instructor Information:**

Instructor Name: Raymond Kessler

MCC Email: rkessler@mclennan.edu

Office Phone Number: 254-299-8184

Office Location: FO 207

Office/Teacher Conference Hours: Tuesday/Thursday: 10:00 am – 11:00, Wednesday: 8:20 am – 9:20 am.

Other Instruction Information: The best way to reach me is through email. I will try to respond to email within 24-36 hours weekdays. It may or may not take longer during weekends or holidays. In person or Zoom meetings can also be scheduled by appointment only.

**Required Text & Materials:**

Title: Chemistry: The Molecular Nature of Matter and Change

Author: Silberberg ~ Amateis

Edition: 9th

Publisher: McGraw-Hill

ISBN: 9781264505463 (E-book, and Connect)

ISBN: 9781265947842 (E-book, Hard copy, and Connect)

McGraw Hill Connect access code. If you buy the textbook from the MCC bookstore, this code is part of the package. Please be aware that used or rented books will NOT have the required access code.

You are required to have your own 'indirectly vented safety goggles' (available in the MCC Bookstore and from other sources).

Other material needed included: nonprogrammable scientific calculator (graphing calculator acceptable), and student research notebook available in the MCC Bookstore.

For online quizzes and/or exams, you will need to download and enable the Lockdown Browser software prior to accessing the assignment. This software is a FREE download. The purpose of this software is to ensure academic integrity. It will prevent you from accessing or opening other applications while completing your course assignments. If a student cannot download the necessary software, it will be the student's responsibility to make arrangements to take online exams and quizzes on another device. Computers are available on campus.

**MCC Bookstore Website:** <http://www.mclennan.edu/bookstore/>

**Methods of Teaching and Learning:**

The course consists of both lecture and laboratory components. Lecture sessions followed by major examinations, outside assignment/checkups, lab exercises with report sheets/checkups and occasional bonus point opportunities. Lecture will be a combination of PowerPoint presentations, video, problem sets, and discussion.

All lessons including PowerPoint slides, quizzes, lab updates, and supplemental videos/homework are available in D2L Brightspace. All homework assignments will be posted in McGraw Hill Connect. Problems sets will require the student be knowledgeable of algebraic manipulation of symbolic equations. It is important that students keep up with material and devote adequate time outside the classroom to study. If any material is unclear, it is recommended to seek guidance as soon as possible. You are responsible for all material covered in the text (unless otherwise stated), even if it is not covered in lecture.

Lab sessions offer hands on experience in a laboratory environment with various activities and resources that will prepare the student to learn the required content for each topic. It is the responsibility of the student to read the lab before the lab session begins to maintain a safe working environment. Students may be required to work independently or as a group. To get credit for lab, the students must turn in a report or complete the activity and submit them on or before their due dates.

**Course Objectives and/or Competencies:**

Lecture:

Upon successful completion of this course, students will:

1. Define the fundamental properties of matter.
2. Classify matter, compounds, and chemical reactions.
3. Determine the basic nuclear and electronic structure of atoms.
4. Identify trends in chemical and physical properties of the elements using the Periodic Table.
5. Describe the bonding in and the shape of simple molecules and ions.
6. Solve stoichiometric problems.
7. Write chemical formulas.
8. Write and balance equations.
9. Use the rules of nomenclature to name chemical compounds.
10. Define the types and characteristics of chemical reactions.
11. Use the gas laws and basics of the Kinetic Molecular Theory to solve gas problems.
12. Determine the role of energy in physical changes and chemical reactions.
13. Convert units of measure and demonstrate dimensional analysis skills.

Laboratory:

Upon successful completion of this course, students will:

1. Use basic apparatus and apply experimental methodologies used in the chemistry laboratory.
2. Demonstrate safe and proper handling of laboratory equipment and chemicals.
3. Conduct basic laboratory experiments with proper laboratory techniques.

4. Make careful and accurate experimental observations.
5. Relate physical observations and measurements to theoretical principles.
6. Interpret laboratory results and experimental data, and reach logical conclusions.
7. Record experimental work completely and accurately in laboratory notebooks and communicate experimental results clearly in written reports.
8. Design fundamental experiments involving principles of chemistry.
9. Identify appropriate sources of information for conducting laboratory experiments involving principles of chemistry

**Course Attendance/Participation Guidelines:**

A complete record of attendance will be maintained for the entire length of the course. Attendance will be taken for laboratory and the lecture. Class attendance is defined as being present in the classroom for at least 75% of the scheduled time period. To meet laboratory attendance, you must be able to complete the experiment and submit the laboratory report sheet during the scheduled laboratory time period. If you arrive tardy and have missed too much of the pre-lab lecture (as determined by the instructor) you may not be allowed to begin the experiment and marked as absent. Students attendance and absences will be reported from the first official day of the class through the last official day of the class. Responsibility for class attendance rests with the student. Regular and punctual attendance is expected for all students.

It has been shown that students who attend class regularly have a higher success rate (read that as passing) than those who chose not to attend class. Each student is responsible for all material presented or assigned and will be held accountable for such materials. Students who stop attending class for any reason, or choose to be withdrawn for any reason, should contact the instructor to officially withdraw from the course. Failure to officially withdraw may result in a failing grade for the course.

If a student is not in attendance in accordance with the policies/guidelines of the class as outlined in the course syllabus as of the course census date, faculty are required to drop students from their class roster prior to certifying the respective class roster. A student's financial aid will be re-evaluated accordingly and the student will only receive funding for those courses attended as of the course census date.

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Before the 60% point of the semester, a student who is absent for 25% or more of a face-to-face or blended course or who misses 25% or more of assigned work for an online course will be withdrawn from the course with a grade of W. A student may also request to be withdrawn with a grade of W before the 60% point of the semester. After the 60% point of the semester, the student may request to be withdrawn if the student is passing, or be assigned the final grade earned at the end of the semester after grades have been updated to reflect missing work.

**Course Outline or Schedule:**

Week	Date	Topic	Assignments Due @ 11:59 pm
1	1/8-1/14	Orientation/Syllabus Chapter 1. Keys to Studying Chemistry	
		<b><i>Lab 0: Orientation/Check in/Safety</i></b>	<b><i>Lab safety</i></b> <i>Due Tue 1/16</i>
2	1/15-1/21	MLK Holiday Chapter 1. Keys to Studying Chemistry	HW 1 <i>Due Sun 1/21</i>
		<b><i>Lab: Used for Lecturing</i></b>	
3	1/22-1/28	Chapter 2. The Components of Matter.	
		<b><i>Lab: Experiment 1. Laboratory Techniques</i></b>	
4	1/29-2/4	Chapter 2. The Components of Matter Chapter 3. Stoichiometry of Formulas and Equations	HW 2 <i>Due Sun 2/4</i>
		<b><i>Lab: Experiment 2. Development of Laboratory Skills and Errors in Measurements</i></b>	
5	2/5-2/11	<b>Exam 1 (Chapters 1-2)</b> Chapter 3. Stoichiometry of Formulas and Equations.	
		<b><i>Lab Part I: Experiment 3. Identification of a Substance Based on Physical Properties</i></b>	
6	2/12-2/18	Chapter 3. Stoichiometry of Formulas and Equations.	HW 3 <i>Due Sun 2/18</i>
		<b><i>Lab Part II: Experiment 3. Identification of a Substance Based on Physical Properties</i></b>	
7	2/19-2/25	Chapter 4. Three Major Classes of Chemical Reactions	
		<b><i>Lab: Experiment 4. Study of Physical and Chemical Changes</i></b>	
8	2/26-3/3	Chapter 4. Three Major Classes of Chemical Reactions	HW 4 <i>Due Sun 3/3</i>

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		<b>Lab: Experiment 5. Determining an Empirical Formula</b>	
	3/4-3/10	<b>Spring Break</b>	
9	3/11-3/17	<b>Exam 2: Chapter 3-4</b> Chapter 5. Gases and the Kinetic-Molecular Theory	
		<b>Experiment 6. Percentage of Water of Hydration</b>	
10	3/18-3/24	Chapter 5. Gases and the Kinetic-Molecular Theory Chapter 6. Thermochemistry Last day for student-initiated withdrawals (3/21)	HW 5 Due Sun 3/24
		<b>Lab 7. Purity of Sodium Bicarbonate</b>	
11	3/25-3/31	Chapter 6. Thermochemistry.	HW 6 Due Sun 3/31
		<b>Lab: Experiment 8. Acid-Bases Titration</b>	
12	4/1-4/7	<b>Exam 3: Chapters 5-6</b> Chapter 7. Quantum Theory and Atomic Structure.	
		<b>Lab: Experiment 9. Physical Behavior of Gases</b>	
13	4/8-4/14	Chapter 7. Quantum Theory and Atomic Structure. Chapter 8: Electron Configuration & Chemical Periodicity	HW 7 Due Sun 4/14
		<b>Lab: Experiment 10. Atomic Mass by Calorimetry</b>	
14	4/15-4/21	Chapter 8: Electron Configuration & Chemical Periodicity	HW 8 Due Sun 4/21
		<b>Lab: Used for Lecturing</b>	
15	4/22-4/28	Chapter 9: Models of Chemical Bonding <b>Exam 4: Chapters 7-8</b>	HW 9 Due Sun 4/28
16	5/2	<b>Final Exam Thursday 11:10 – 1:10 pm</b>	

HW: Homework; Lab: Lab reports are due at the beginning of the next lab period

**This schedule is subject to change.** You will be informed of any changes to the syllabus by email, D2L Brightspace, or class announcements.

### **Course Grading Information:**

Your course grade will be based on the points received from the exams, final, lab grades, quizzes, and homework assignments.

**Exam:** Exams (4) will count for 40% of the course grade. There will be an exam at the end of the discussion of each unit covered. Exams will be given in class and must be completed during one regular class period. Work must be legible, and the final answer must be clearly indicated to



receive credit. If special accommodations become necessary, arrangements must be made BEFORE the exam date. There are no makeup exams without prior arrangements. Students arriving late to class may not start an exam once the first person completes the exam and leaves the classroom, and additional time will not be given. You may not leave the room during a test. During tests, there are no cell phones, or any other electronic devices allowed. You must use the periodic tables, information sheets, and scratch paper supplied by the department. All tests will be comprehensive and may contain true/false, multiple choice, matching, completion, and short answer questions, as well as problem solving when applicable.

The final exam is comprehensive, and there are no exemptions from taking the final. It counts for 15% of the course grade. If you are a graduating student, you MUST take the final exam on Monday or Tuesday the week of finals. There will be no make-up test for the final exam. Your non-curved Final Exam grade will replace your lowest unit exam grade, excluding a grade of zero (0), if your final exam grade is higher than your lowest unit exam.

**Pre-lecture quizzes/assignment:** Pre-lecture assignments will be given before the topic is discussed in class to encourage students to read the material before presented. It represents 10% of your course grade. Pre-lecture assignments will be assigned for each chapter on the McGraw Hill Connect online platform. Pre-lecture assignments may cover problems/exercises assigned during previous class sessions. You are expected to complete the assignment by the due date. No pre-lecture assignment will be accepted late without prior accommodations.

**Homework:** Homework will be assigned for each chapter on the McGraw Hill Connect online platform. The homework assignment is designed to encourage students to keep up with the material, and master concepts evaluated on the exams. Homework represents 15% of your final course grade. You are expected to comply with the MCC academic Integrity Statement when completing homework assignments. Homework is due on Sundays at 11:59 CST after completion of the chapter. No homework assignment will be accepted late without prior accommodations.

**Lab Grades:** The lab portion will make up 20% of the final course grade. You are encouraged to keep good report (answer) sheets during laboratory session. Lab reports are due the following week. Labs cannot be made-up, but the lowest lab grade will be dropped. Failure to attend a lab will result in a lab grade of zero and a missed lab grade will not be dropped unless prior arrangements to miss that lab have been made.

Each laboratory grade will consist of two portions: Lab report and a copy of the lab notebook.

**Lab reports and research notebook** must be in permanent ink. Felt-tip pens, erasable ink, and pencils are not allowed. A minimum of 10 points is deducted for failure to follow these guidelines. One point is deducted for each error in significant figures or units. Changes to data may only be made by drawing one line through the information to be changed. One point is deducted for each instance in which a student writes over or obscures in any way, previously recorded data. Do NOT use correction fluid or tape. The lab report must be turned in at the beginning of the next lab. Failure to comply with safety and other requirements can reduce a grade significantly, or result in a student's receiving a grade of zero for the day's lab.

The use of the research notebook in the lab is MANDATORY. It is part of the lab grade. Data and observations must be written directly into your research notebook in black pen. Do NOT use the lab report sheet to collect data and observations. You CANNOT write data (such as weights, measurements, etc.) on scratch paper. Failure to follow the above rules will result in a grade of zero for that lab's day. **You must show your research notebook at the end of each lab session to be signed by your instructor.** Your lab notebook must include the following information:

- 1) Up-to-date Table of Contents
- 2) Pre-lab lecture notes. This section should contain the date, the name of the experiment, and your notes from the pre-lab. Contents should include:
  - a) Safety instructions given in pre-lab lecture
  - b) Any changes from the directions in the lab textbook
  - c) Additional information providing background on chemicals, etc.
  - d) Calculations methods
- 3) Data, observations, calculations, etc.

In a laboratory setting, safety equipment must be worn at all times: long pants/skirt (covering at least the top half of the calf), apron or lab coat, hair back, safety goggles, and, if necessary, gloves. No open-toed shoes, shoes with holes in them, shoes that leave the top of the foot exposed, hats of any sort, shorts, food or drink are allowed. If a student is not dressed in the appropriate attire, that student cannot attend lab until the requirements are met. Any student who is not able to attend lab due to inappropriate attire will receive a zero grade for that assignment with no make-up or dropped grade for that assignment. Anyone acting in an unsafe manner will be warned once. If seen without safety equipment or acting improperly a second time, they will be asked to leave the laboratory. They will be allowed to return in 30 minutes to finish their work, if they can. If they are asked to leave more than once for any given experiment, they will receive a zero for that experiment's lab report. Safety is the MOST important part of lab. The use of goggles in lab is mandatory. Food or drinks are not allowed in the laboratory. Students not following these regulations will not be admitted into lab under any circumstances and will

receive a grade of zero for that lab. Students must abide by the general safety regulations as described.

**Overall Grade:** It is the students' responsibility to keep track of their overall grade based on Quizzes - 10%, Homework – 15%, Exams – 40%, Lab – 20%, and Final 15%. If you have trouble calculating your grade, you may ask to see your overall grade. Final grades are based on 90.00+ = A, 80.00-89.99 = B, 70.00-79.99 = C, 60.00-69.99 = D, and less than 60.00 = F. Final Grades will not be rounded up, Grades will not be released over the phone or by email. Work that is not easily readable or without a name will not be graded. If you wish to dispute your grade, please contact the instructor within one week.

All students who enroll at MCC are admitted with the expectation that they will demonstrate integrity in every aspect of their work both for and with other members of this academic community. Cheating of any kind will not be tolerated. If there is any evidence of cheating (obtaining or attempting to obtain, or aiding another to obtain credit for work, or any improvement in evaluation of performance, by any dishonest or deceptive means) or plagiarizing (taking on someone else's work idea and passing them off as your own) on any homework, quiz, test, or final exam, you will receive a zero for that item and cannot make it up or replace it and it cannot be dropped. The result of a second offence will be a failing grade in the course. Student Discipline/Student Development may take additional action.

Anyone caught cheating or plagiarizing on any assignment will have grounds to fail the course for the semester and be reported to the disciplinary council. Cheating may include, but not limited to sharing assignments or test answers, collusion, having someone else do your assignment, posting information for the class on websites like Chegg, Slate and other similar sites.

**Late Work and Make Up Work Policies:**

Connect McGraw-Hill assignments, laboratory assignments, and quizzes are not accepted late. A make-up exams/quizzes will be given only with prior arrangements and must be taken within one week upon your return to class. Missed quiz/exam is recorded as a zero and may not be made up unless prior arrangements have been made. Documentation of the extenuating circumstances will be required. If you arrive tardy and have missed too much of the pre-lab lecture (as determined by the instructor) you may not be allowed to begin the experiment. Please, arrive ON TIME to avoid a zero grade. If you do not complete the lab, you cannot turn in a lab report, and receive a

zero grade for that lab. Students are responsible for checking the due date of all assignments. While I encourage you to come to me with questions about answers on assignments anytime, if you wish to dispute a grade or have a grade changed, you must contact me about it within one weeks of the assignment due date or there will be no change to the grade.

Students will be permitted to make up class work and assignments missed due to absences caused by (1) authorized participation in official College functions, (2) personal illness, (3) an illness or a death in the immediate family or (4) Pregnant or Parenting Protections under the Title IX and/or Texas State Education Code or (5) the observance of a religious holy day (6) or military service. Also, the instructor has the right of determining whether a student may make up work missed due to absences for other reasons. It is the student's responsibility to inform the instructor of the reason for an absence and to do so in a timely fashion.

Students, whether present or absent, are responsible for all material presented or assigned for the course and will be held accountable for such materials in the determination of course grades. The student is responsible for any work missed regardless of the cause of the absence. The student must discuss such work with the instructor and should do so in a timely fashion. Communication between the student and instructor is important and it is the student's responsibility to initiate such communication.

**Student Behavioral Expectations or Conduct Policy:**

*"Students are expected to maintain classroom decorum that includes respect for other students and the instructor, prompt and regular attendance, and an attitude that seeks to take full advantage of the education opportunity."*

Each student is expected to behave in a civil and respectful manner toward the instructor and other students in all forms of communication. Infractions will not be tolerated. Failure to comply will be grounds for dismissal from the class and name submitted to Students Discipline.

If there is any evidence of cheating on any homework, quiz, test, or final exam, you will receive a zero for that item and cannot make it up or replace it and it cannot be dropped. The result of a second offence will be failing grade in the course. Furthermore, you could also face expulsion from MCC.

Safety equipment must be worn at all times: long pants/skirt (covering at least the top half of the calf), apron or lab coat, hair back, safety goggles, and, if necessary, gloves. No open-toed shoes, shoes with holes in them, shoes that leave the top of the foot exposed, hats of any sort, shorts, food or drink are allowed. Anyone acting in an unsafe manner will be warned once. If seen without safety equipment or acting improperly a second time, they will be asked to leave the laboratory. If they are asked to leave more than once for any given experiment, they will receive a zero for that experiment's lab report. Safety is the MOST important part of lab. Students must abide by the general safety regulations as described in the chemistry laboratory safety rules.

**[Click Here for the MCC Attendance/Absences Policy](https://www.mclennan.edu/highlander-guide/policies.html)**

**(<https://www.mclennan.edu/highlander-guide/policies.html>)**

Click on the link above for the college policies on attendance and absences. Your instructor may have additional guidelines specific to this course.

10/09/2023



## **ACADEMIC RESOURCES/POLICIES**

### **Accommodations/ADA Statement:**

Any student who is a qualified individual with a disability may request reasonable accommodations to assist with providing equal access to educational opportunities. Students should contact the Accommodations Coordinator as soon as possible to provide documentation and make necessary arrangements. Once that process is completed, appropriate verification will be provided to the student and instructor. Please note that instructors are not required to provide classroom accommodations to students until appropriate verification has been provided by the Accommodations Coordinator. For additional information, please visit [www.mclennan.edu/disability](http://www.mclennan.edu/disability)

Students with questions or who require assistance with disabilities involving physical, classroom, or testing accommodations should contact:

[disabilities@mclennan.edu](mailto:disabilities@mclennan.edu) (254)299-8122  
Room 319, Student Services Center

### **Title IX:**

We care about your safety, and value an environment where students and instructors can successfully teach and learn together. If you or someone you know experiences unwelcomed behavior, we are here to help. Individuals who would like to report an incident of sexual misconduct are encouraged to immediately contact the Title IX Coordinator at [titleix@mclennan.edu](mailto:titleix@mclennan.edu) or by calling, Dr. Claudette Jackson, (Accommodations/Title IX) at (254) 299-8465. MCC employees are mandatory reporters and must report incidents immediately to the Title IX Coordinator. Individuals may also contact the MCC Police Department at (254) 299-8911 or the MCC Student Counseling Center at (254) 299-8210. The MCC Student Counseling Center is a confidential resource for students. Any student may report sexual harassment anonymously by visiting <http://www.lighthouse-services.com/mclennan/>

Additionally, Title IX provides rights and protections for pregnant and newly parenting students. Go to McLennan's Title IX webpage at [www.mclennan.edu/titleix/](http://www.mclennan.edu/titleix/). It contains more information about definitions, reporting, confidentiality, resources, and what to do if you or someone you know is a victim of sexual misconduct, gender-based violence or the crimes of rape, acquaintance rape, sexual assault, sexual harassment, stalking, dating violence, or domestic violence.

**Student Support/Resources:**

MCC provides a variety of services to support student success in the classroom and in your academic pursuits to include counseling, tutors, technology help desk, advising, financial aid, etc. A listing of these and the many other services available to our students is available at <http://www.mclennan.edu/campus-resource-guide/>

Academic Support and Tutoring is here to help students with all their course-related needs. Specializing in one-on-one tutoring, developing study skills, and effectively writing essays. Academic Support and Tutoring can be found in the Library and main floor of the Learning Commons. This service is available to students in person or through Zoom. You can contact the Academic Support and Tutoring team via Zoom or email ([ast@mclennan.edu](mailto:ast@mclennan.edu)) by going to our website (<https://www.mclennan.edu/academic-support-and-tutoring/>)

College personnel recognize that food, housing, and transportation are essential for student success. If you are having trouble securing these resources or want to explore strategies for balancing life and school, we encourage you to contact either MCC CREW – Campus Resources Education Web by calling (254) 299-8561 or by emailing [crew@mclennan.edu](mailto:crew@mclennan.edu) or a Success Coach by calling (254) 299-8226 or emailing [success@mclennan.edu](mailto:success@mclennan.edu).

Paulanne's Pantry (MCC's food pantry) provides free food by appointment to students, faculty and staff. To schedule an appointment, go to <https://calendly.com/paulannespantry-mcc/15min>.

The CREW, Success Coaches, and Paulanne's Pantry are all located on the second floor of the Student Services building in Success Coaching Services.

**MCC Foundation Emergency Grant Fund:**

Unanticipated expenses, such as car repairs, medical bills, housing, or job loss can affect us all. Should an unexpected expense arise, the MCC Foundation has an

emergency grant fund that may be able to assist you. Please go to <https://www.mclennan.edu/foundation/scholarships-and-resources/emergencygrant.html> to find out more about the emergency grant. The application can be found at [https://www.mclennan.edu/foundation/docs/Emergency\\_Grant\\_Application.pdf](https://www.mclennan.edu/foundation/docs/Emergency_Grant_Application.pdf)

**MCC Academic Integrity Statement:**

Please view our [Academic integrity statement](#) for more information about academic integrity, dishonesty, and cheating. The unauthorized use of artificial intelligence (AI) for classwork can be a violation of the College's General Conduct Policy. Whether AI is authorized in a course and the parameters in which AI can be used in a course will be outlined by each instructor.

**Minimum System Requirements to Utilize MCC's D2L|Brightspace:**

Go to <https://www.mclennan.edu/center-for-teaching-and-learning/Faculty-and-Staff-Commons/requirements.html> for information on the minimum system requirements needed to reliably access your courses in MCC's D2L|Brightspace learning management system.

**Minimum Technical Skills:**

Students should have basic computer skills, knowledge of word processing software, and a basic understanding of how to use search engines and common web browsers.

**Backup Plan for Technology:**

In the event MCC's technology systems are down, you will be notified via your MCC student email address. Please note that all assignments and activities will be due on the date specified in the Instructor Plan, unless otherwise noted by the instructor.

**Email Policy:**

McLennan Community College would like to remind you of the policy (<http://www.mclennan.edu/employees/policy-manual/docs/E-XXXI-B.pdf>) regarding college email. All students, faculty, and staff are encouraged to use their McLennan email addresses when conducting college business.

A student's McLennan email address is the preferred email address that college employees should use for official college information or business. Students are expected to read and, if needed, respond in a timely manner to college emails. For more information about your student email account, go to [www.mclennan.edu/studentemail](http://www.mclennan.edu/studentemail).



**Instructional Uses of Email:**

Faculty members can determine classroom use of email or electronic communications. Faculty should expect and encourage students to check the college email on a regular basis. Faculty should inform students in the course syllabus if another communication method is to be used and of any special or unusual expectations for electronic communications.

If a faculty member prefers not to communicate by email with their students, it should be reflected in the course syllabus and information should be provided for the preferred form of communication.

**Email on Mobile Devices:**

The College recommends that you set up your mobile device to receive McLennan emails. If you need assistance with set-up, you may email [Helpdesk@mclennan.edu](mailto:Helpdesk@mclennan.edu) for help.

You can find help on the McLennan website about connecting your McLennan email account to your mobile device:

- [Email Setup for iPhones and iPads](#)
- [Email Setup for Androids](#)

**Forwarding Emails:**

You may forward emails that come to your McLennan address to alternate email addresses; however, the College will not be held responsible for emails forwarded to an alternate address that may be lost or placed in junk or spam filters.

For more helpful information about technology at MCC, go to [MCC's Tech Support](#) or email [helpdesk@mclennan.edu](mailto:helpdesk@mclennan.edu).

**Disclaimer:**

The resources and policies listed above are merely for informational purposes and are subject to change without notice or obligation. The College reserves the right to change policies and other requirements in compliance with State and Federal laws. The provisions of this document do not constitute a contract.